# Notes

## Acetylated Sesterterpenes from the Great Barrier Reef Sponge Luffariella variabilis<sup>§</sup>

Piers Ettinger-Epstein,<sup>†,‡</sup> Cherie A. Motti,<sup>†</sup> Rocky de Nys,<sup>‡</sup> Anthony D. Wright,<sup>†</sup> Christopher N. Battershill,<sup>†</sup> and Dianne M. Tapiolas<sup>\*,†</sup>

Australian Institute of Marine Science PMB 3, Townsville MC, Queensland, 4810, Australia, School of Marine Biology and Aquaculture, James Cook University, Townsville, Queensland, 4811, Australia, and AIMS@JCU Sir George Fisher Building, James Cook University, Townsville, Queensland, 4811, Australia

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Chemical investigation of the sponge *Luffariella variabilis* collected from the Palm Island group of the Great Barrier Reef, Australia, yielded three new acetylated compounds, 25-acetoxyluffariellin A (1), 25-acetoxyluffariellin B (2), and 25-acetoxyseco-manoalide (3). The structures of the new compounds were elucidated on the basis of interpretation of their spectroscopic data. The known metabolites manoalide (4), seco-manoalide (5), luffariellin A (8), and manoalide monoacetate (10) were also isolated. The new acetylated compounds (1–3) were labile in the sponge tissue when samples were allowed to thaw prior to extraction, but were stable once isolated. Sponge samples that were completely thawed contained only hydroxylated compounds (alcohols). This finding supported the deduction that the acetylated compounds are being enzymatically transformed and/or degraded.

Sponges of the genus Luffariella are widespread throughout the Indo-Pacific and have afforded a wealth of bioactive sesterterpenes.<sup>1–5</sup> Manoalide (4) was the first of a series of related compounds reported from the Palauan sponge Luffariella variabilis by De Silva and Scheuer,<sup>1</sup> who subsequently isolated seco-manoalide (5) and (E)and (Z)-neomanoalide (6, 7).<sup>2</sup> Kernan and Faulkner<sup>3</sup> reported the presence of two new sesterterpenes, luffariellin A (8) and luffariellin B (9), in addition to manoalide (4) and seco-manoalide (5). They also quantified variation of chemistry in 410 Palauan sponges all assigned as L. variabilis, with the ratio of these four metabolites being found to vary significantly between different sponge samples.<sup>3</sup> In the current study the isolation of three new acetylated compounds (1-3) and the previously reported manoalide (4),<sup>1</sup> seco-manoalide (5),<sup>2</sup> luffariellin A (8),<sup>3</sup> and manoalide monoacetate  $(10)^6$  is reported from L. variabilis, collected from the Palm Island Group of the Great Barrier Reef, Australia. Furthermore, it has been found that sponge storage protocols have a significant effect on the isolated chemistry. This report details the isolation and structure elucidation of three new compounds and the effect of allowing samples to thaw on the presence of these secondary metabolites. The sponge L. variabilis collected off Orpheus Island, Australia, was frozen as soon as returned to the surface by immersion in liquid N<sub>2</sub> and kept frozen at -176 °C. The frozen sponge was freeze-dried and extracted with  $CH_2Cl_2$  (3  $\times$  200 mL) at RT. The combined dichloromethane extracts (358 mg) of L. variabilis were purified using a series of preparative and semipreparative HPLC chromatography. This process yielded three new sesterterpenes, 1-3, as well as the known compounds manoalide (4),<sup>1</sup> seco-manoalide (5),<sup>2</sup> luffariellin A (8),<sup>3</sup> and manoalide monoacetate (10).<sup>6</sup> A number of modifications to previous structural assignments of the known metabolites were recorded during the elucidation of these compounds. The previous isolation of luffariellin A reported doubled proton and carbon signals in the NMR spectra in purified CDCl<sub>3</sub> for the protons and carbons around the  $\alpha$ -hydroxybutenolide and

 $\delta$ -lactol rings. This was consistent with a mixture of two diastereoisomers in these regions. However, in slightly acidic CDCl<sub>3</sub>, only a single set of broad signals were observed. This doubling of signals has also been reported for manoalide<sup>7</sup> and related compounds.<sup>8</sup> We also observed a broadening of signals in the NMR spectra for the compounds reported herein.

Compound 1 showed a  $[M + Na]^+$  ion in its HRESIMS, consistent with the molecular formula C27H38O6 and, therefore, nine degrees of unsaturation in the form of multiple bonds and rings. The <sup>1</sup>H and <sup>13</sup>C NMR data of 1 showed it to contain six double bonds and therefore be tricyclic and, thus, very similar to luffariellin A (8).3 When the <sup>1</sup>H NMR data of 1 and 8 were compared, the differences between the two data sets were the presence of an additional methyl singlet signal at  $\delta_{\rm H}$  2.18 and the downfield shift of the H-25 resonance ( $\delta_{\rm H}$  7.12 in 1 compared to  $\delta_{\rm H}$  6.22 in luffariellin A). The <sup>13</sup>C NMR spectral data of 1 compared to that of luffariellin A (8) showed additional carbon signals at 168.8 (qC) and 20.1 (CH<sub>3</sub>) ppm, and the signals associated with C-3 and C-25 were shifted ( $\delta_{\rm C}$  165.5, 92.4 in **1** compared to  $\delta_{\rm C}$  169.0/168.3, 98.3 in luffariellin A). These differences were consistent with 1 being the 25-acetoxyl derivative of luffariellin A (8). The gHMBC NMR data of 1 confirmed this and confirmed the position of the acetoxyl function. A correlation from the methyl signal at  $\delta_{\rm H}$  2.18 to C-25  $(\delta_{\rm C} 92.4)$  was observed as well as correlations from H-25  $(\delta_{\rm H} 7.12)$ to C-2 ( $\delta_{\rm C}$  118.6) and C-3 ( $\delta_{\rm C}$  165.5). These correlations are in agreement with the NMR data for both manoalide monoacetate (10) and thorectolide monoacetate (11), both of which have an acetoxybutenolide terminus.<sup>8</sup> The  $\Delta^{10}$  geometry was determined to be E on the basis of the <sup>13</sup>C NMR chemical shift of C-23 ( $\delta_{\rm C}$  16.2).<sup>9</sup> Compound 1 is therefore 25-acetoxyluffariellin A. Selected  ${}^{13}C$ NMR assignments and relative configurations were confirmed from the gCOSY, gHSQC, and 1D selective TOCSY spectra of 1. The chemical shifts of C-13 and C-18 resonated at  $\delta_{\rm C}$  34.2 and 28.8, respectively, in contrast to the values previously reported<sup>3</sup> (C-13 and C-18 at  $\delta_{\rm C}$  25.9 and 34.3, respectively). The stereochemistry at C-14 relative to C-15 was determined by a 1D selective gNOESY experiment. When the H<sub>3</sub>-22 signal ( $\delta_{\rm H}$  0.72) was irradiated, NOEs were observed to the H<sub>3</sub>-21 signal ( $\delta_{\rm H}$  1.68), the signal for the adjacent methine H-15 ( $\delta_{\rm H}$  1.77), and to both of the signals

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<sup>\*</sup> Corresponding author. Tel: +61 7 4753 4452. Fax: +61 7 4772 5852. E-mail: d.tapiolas@aims.gov.au.

<sup>&</sup>lt;sup>†</sup> Australian Institute of Marine Science.

<sup>&</sup>lt;sup>‡</sup> James Cook University.

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associated with the exo-methylene H<sub>2</sub>-20 ( $\delta_{\rm H}$  4.84, 4.65), indicating these groupings to be on the same side of the ring, which gives **1** the same relative stereochemistry at C-14 and C-15 as that previously reported.<sup>3</sup> We were unable to conclusively determine the relative stereochemistry at C-4 or C-25. The axial nature of H-4 was deduced from its coupling constants (10.9, 4.0 Hz). When the H-4 signal ( $\delta_{\rm H}$  4.77) was irradiated, small NOEs were observed to the signals for H-2, H<sub>2</sub>-5, H-6, H-24, H-25, and H<sub>3</sub>-25-OAc as well to the signal for the exchangeable proton 4-OH. Consequently the relative stereochemistry at both C-4 and C-25 remains undetermined.

Compound **2** analyzed for  $C_{27}H_{38}O_6$  by HRESIMS. Comparison of the NMR data of compound **2** (Tables 1 and 2) with those of luffariellin B (**9**)<sup>3</sup> showed the two data sets to be very similar. Where differences were apparent, they were consistent with **2** being the C-25 acetoxyl derivative of luffariellin B (**9**). These differences included additional signals for an acetate group [ $\delta_C$  169.1 (qC), 20.6 (CH<sub>3</sub>),  $\delta_H$  2.21 (s)] and the shift of the C-3, C-25, and H-25 signals ( $\delta_C$  166.1, 93.2 and  $\delta_H$  7.21 in **2** compared to  $\delta_C$  170.4/ 169.3, 98.3/97.9 and  $\delta_H$  5.40 in luffariellin B). These differences showed **2** to be 25-acetoxyluffariellin B. Similar to luffariellin A, the chemical shifts of C-13 and C-18 for luffariellin B<sup>3</sup> required revision. The relative stereochemistry of C-14 and C-15 was determined by selective gNOESY experiments to be the same as that reported for luffariellin A.<sup>3</sup> The relative stereochemistry at C-4 and C-25 remains unassigned.

Compound **3** had the molecular formula  $C_{27}H_{38}O_6$ , as determined by HRESIMS, and thus was isomeric with compounds **1** and **2**. Comparison of the 1D NMR data of compound **3** (Tables 1 and 2) with that of seco-manoalide (**5**)<sup>2</sup> showed the two data sets to be similar. In the data set for **3** additional resonances in both the <sup>1</sup>H and <sup>13</sup>C NMR spectra [ $\delta_{\rm H}$  2.20 (s);  $\delta_{\rm C}$  169.1 (qC), 20.7 (CH<sub>3</sub>)] and a shift in the signals assigned as C-3, C-25, and H-25 ( $\delta_{\rm C}$  165.9, 93.2;  $\delta_{\rm H}$  7.07 respectively in **3**, compared to  $\delta_{\rm C}$  169.0, 99.0;  $\delta_{\rm H}$  6.15 respectively in seco-manoalide) were consistent with the presence of an acetoxyl function at C-25 in **3** rather than the OH of seco-manoalide. As for **1** and **2**, the geometry of the C-10 olefin was determined to be *E* on the basis of the chemical shift of C-23 ( $\delta_{\rm C}$  15.7). These data and all of the other physical data recorded were consistent with **3** being 25-acetoxyseco-manoalide. The relative stereochemistry at C-4 and C-25 remains unassigned.

Optical rotations measured for manoalide (4) and manoalide monoacetate (10) in this study agree with published values,<sup>1,6</sup> indicating that the absolute stereochemistry at C-4 in both can be assigned as R. The optical rotation of seco-manoalide isolated in this study ( $[\alpha]_D$  –65.3) differed in magnitude and sign from that reported in the literature ( $[\alpha]_D$  +16.2).<sup>2</sup> However, because both compounds are mixtures of diastereoisomers at C-25, it is not possible to comment on the significance of this finding or the absolute stereochemistry of seco-manoalide. The optical rotation measured for luffariellin A ( $[\alpha]_D$  –32.0) was of similar magnitude and opposite to that reported in the literature ([ $\alpha$ ]<sub>D</sub> +40.1).<sup>3</sup> The relative stereochemistry at C-14 and C-15 in luffariellin A isolated in the current investigation was shown by selective gNOESY experiments to be the same as reported.<sup>3</sup> Due to both compounds being isolated as mixtures of diastereoisomers, it is not possible to confirm unequivocally the current compound as the antipode of that previously reported.3

The presence of the new acetylated compounds (1-3) in the sponge extracts was significantly affected by the sponge tissue

 $\delta_{\rm H} \left( J \text{ in Hz} \right)$ 

4.77 (dd, 10.9, 4.0)

5.71 (d, 4.3)

5.10 (dt, 1.0, 6.7)

2.14 (m)

2.15 (m)

1.71 (m) 1.76 (m)

1.36 (m) 1.40 (m)

1.77 (m)

1.30 (m)

1.94(m)

1.61 (m)

1.72 (m)

1.48 (m)

1.75 (m)

4.65 (s)

4.84 (s)

1.68 (s)

1.60 (s)

7.12 (s)

2.18(s)

0.72 (d, 7.0)

5.31 (d, 4.5)

2.82 (d, 4.5)

2.28 (ddd 17.1, 4.3, 4.0) 2.32 (dd 17.1, 10.9)

6.10 (s)

position

1 2

3 4

6

7 8

9

10

11 12

13

14 15

16

17

18

19

20

21

22

23

24

25

24-OH

25-OAc

4-OH 5 1

Table 1. <sup>1</sup>H NMR Spectroscopic Data (600 MHz, CDCl<sub>3</sub>) for Compounds 1-3

1.76 (m)

1.32 (m)

1.94(m)

1.61 (m)

1.72 (m)

1.46 (m)

1.72 (m)

4.64 (s)

4.83 (s)

1.68 (s)

1.56(s)

9.44 (s)

7.07 (s)

2.21 (s)

0.71 (d, 7.0)

14, 19, 21

14, 19, 21

16

6,8

2,3

10.12

	2		3	
gHMBC	$\delta_{\rm H} (J \text{ in Hz})$	gHMBC	$\delta_{\rm H} (J \text{ in Hz})$	gHMBC
1, 3, 4, 25	6.15 (s)	1, 3, 4, 25	6.15 (s)	1, 3, 4, 25
2, 3, 25	4.68 (dd, 7.0, 4.5, 4.0) 2.42 (d, 4.5)	2, 3, 5, 6	4.68 (dd, 7.1, 5.2, 4.0) 2.40 (d, 5.2)	2, 3, 5, 6,
3, 4, 6, 7	2.81 (dt, 15.4, 7.1) 2.89 (ddd, 15.4, 7.1, 4.0)	4, 6	2.82 (dt, 15.4, 7.1) 2.91 (ddd, 15.4, 7.1, 4.0)	4, 6
4, 5, 8, 24	6.53 (t, 7.1)	4, 5, 7, 8, 24	6.55 (t, 7.1)	4, 5, 7, 8, 24
	2.32 (t, 7.6)	7, 9, 10, 24	2.35 (t, 7.6)	7, 9, 10, 24
	2.08 (m)	8, 10, 11	2.11 (m)	8, 10, 11
3, 9, 12, 23	5.08(br t, 7.2)	8, 9, 12, 13, 23	5.13 (br t, 7.2)	8, 9, 12, 23
	1.68 (m)		1.98 (m)	
	1.74 (m)		· ·	
	1.32 (m)		2.01 (m)	
	1.38 (m)			

14, 19, 21

14, 19, 21

16

10, 12

6,8

2,3

Fable 2	<sup>13</sup> C NMR Spectroscopic Data (125 MHz CDCl <sub>2</sub> ) for	
	1 - 1 - 2	
Compoun	ds 1-3	

	1	2	3
position	$\delta_{\rm C}$ , mult.	$\delta_{\rm C}$ , mult.	$\delta_{\rm C}$ , mult.
1	169.2, qC	168.7, qC	168.6, qC
2	118.6, ĈH	119.5, CH	119.8, ĈH
3	165.5, qC	166.1, qC	165.9, qC
4	61.5, CH	65.9, CH	65.8, CH
5	27.9, CH <sub>2</sub>	34.2, CH <sub>2</sub>	33.9, CH <sub>2</sub>
6	120.6, CH	146.6, CH	146.2, CH
7	136.9, CH	146.2, qC	146.5, qC
8	32.3, CH <sub>2</sub>	24.3, CH <sub>2</sub>	24.4, CH <sub>2</sub>
9	25.6, CH <sub>2</sub>	26.4, CH <sub>2</sub>	26.7, CH <sub>2</sub>
10	122.6, CH	122.1, CH	121.9, CH
11	136.4, qC	137.7, qC	137.7, qC
12	34.7, CH <sub>2</sub>	34.8, CH <sub>2</sub>	40.1, CH <sub>2</sub>
13	34.2, CH <sub>2</sub>	34.2, CH <sub>2</sub>	27.8, CH <sub>2</sub>
14	55.1, qC	55.1, qC	136.9, qC
15	41.8, CH	41.9, CH	127.1, qC
16	30.6, CH <sub>2</sub>	31.0, CH <sub>2</sub>	32.7, CH
17	20.2, CH <sub>2</sub>	20.6, CH <sub>2</sub>	19.5, CH <sub>2</sub>
18	28.8, CH <sub>2</sub>	29.5, CH <sub>2</sub>	39.8, CH <sub>2</sub>
19	148.0, qC	148.1, qC	35.0, qC
20	111.6, CH <sub>2</sub>	111.8, CH <sub>2</sub>	28.6, CH <sub>3</sub>
21	20.2, CH <sub>3</sub>	20.7, CH <sub>3</sub>	28.6, CH <sub>3</sub>
22	17.7, CH <sub>3</sub>	18.2, CH <sub>3</sub>	19.8, CH <sub>3</sub>
23	16.2, CH <sub>3</sub>	16.3, CH <sub>3</sub>	15.7, CH <sub>3</sub>
24	91.4, CH	194.6, qC	194.2, qC
25	92.4, CH	93.2, CH	93.2, CH
25-OAc	168.8, qC	169.1, qC	169.1, qC
	20 0.1, CH <sub>3</sub>	20.6, CH <sub>3</sub>	20.7, CH <sub>3</sub>

workup protocols. A small number of sponge samples, which were known to contain the acetylated compounds (1-3 and 10), were subsampled and thawed before being lyophilized and extracted. Only the nonacetylated compounds (4, 5, and 8) were isolated from

the CH<sub>2</sub>Cl<sub>2</sub> extracts of these samples. Subsequently, sponge samples (n = 15) were divided into two *in situ* and immediately returned to the surface, where they were immersed in liquid N2. In the laboratory, one piece of each collected sponge was allowed to thaw for 3 h prior to being freeze-dried, with the remaining piece kept frozen until freeze-dried. In the sponge samples that were kept frozen before being freeze-dried and extracted, the acetvlated compounds (1-3 and 10) together with compounds 4, 5, and 8 were isolated. However, in the sponge samples that were allowed to thaw before being freeze-dried and extracted, only the nonacetylated compounds 4, 5, and 8 were detected. Once isolated, the acetylated compounds 1-3 and 10 are stable, suggesting that their "instability" in the sponges is enzyme mediated. Presumably the enzymes that are active in the freshly collected sponge remain viable in the frozen material. As such, the acetylated compounds may be precursor storage metabolites that can be hydrolyzed enzymatically to the alcohols that have a predetermined function, for example, defense. Activated defenses where biologically inactive acetylated metabolites are enymatically hydrolyzed to biologically active alcohols or aldehydes have been previously reported in the algae Halimeda spp.<sup>10</sup> and Caulerpa taxifolia, C. prolifera, and C. racemosa.<sup>11</sup> Recent reports<sup>12</sup> have also postulated enzymatic cleavage of brominated isoxazoline alkaloids into more active monocylic nitrogenous compounds (aeroplysinin-1 and dienones) as an activated defense mechanism after mechanical wounding in the sponge Aplysina sp.

1.91 (t, 6.2)

1.56 (m)

1.42 (m)

0.99 (s)

0.99 (s)

1.60 (s)

1.62 (s)

9.45 (s)

7.07 (s)

2.20 (s)

18

18

16

6,8

2

10, 12

### **Experimental Section**

**General Experimental Procedures.** Optical rotations were recorded on a Jasco 715 CD polarimeter. UV spectra were measured on a Shimadzu SPD-M10AVP PDA detector. Infrared spectra were taken on a Nicolet Nexus FTIR. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded in neutralized CDCl<sub>3</sub> using a Bruker Avance 600 MHz NMR spectrometer

with cryoprobe. Spectra were referenced to residual <sup>1</sup>H ( $\delta$  7.27) and  $^{13}C$  ( $\delta$  77.0) resonances in the deuterated solvents. Both 1D and 2D NMR spectra were recorded using standard Bruker pulse sequences. High-resolution mass spectra were measured with a Bruker BioApex 47e FT-ICR mass spectrometer fitted with an Analytica of Branford electrospray source. Ions were detected in positive mode within a mass range of m/z 200–1000. Direct infusion of the sample (0.2 mg mL<sup>-1</sup>) was carried out using a Cole Palmer 74900 syringe pump at a flow rate of 80 µL h<sup>-1</sup>. HPLC was performed with a Shimadzu LC10-AT pump coupled to either a SPD-M10AVP PDA detector (analytical analyses) or a Shimadzu SPD-10A UV/vis detector (preparative isolations). HPLC columns were purchased from Phenomenex. Compressed gases came from BOC Gases (Townsville, Australia) and were at least 99.99% pure. Purified water was obtained from a MilliQ water purification system (Millipore, Billerica, MA); all other solvents used were HPLC grade (Mallinckrodt, Hazelwood, MO).

**Sponge Material.** *Luffariella variabilis* (order Dictyoceratida, family Thorectidae) was collected by hand using scuba at depths between 5 and 10 m off Orpheus Island, Australia. A voucher specimen is lodged at the Australian Institute of Marine Science, Queensland, Australia (#27405). Freshly collected sponges were frozen as soon as returned to the surface by immersion in liquid N<sub>2</sub> and kept frozen at -176 °C. The sponges had a dark brown-black exterior and were extensively covered in fouling organisms; interiors were orange-brown.

Extraction and Isolation. The frozen sponge sample for preparative isolation was freeze-dried and extracted with  $CH_2Cl_2$  (3 × 200 mL) at RT. The combined dried extracts (358 mg) were dissolved in MeOH and chromatographed using preparative RP HPLC [Phenomenex, Luna C18 (2), 5  $\mu$ m, 250  $\times$  21 mm; gradient of CH<sub>3</sub>CN-H<sub>2</sub>O, 70:30 to 100:0, over 60 min as eluent, flow rate 10 mL/min; UV detection at 254 nm]. A late eluting fraction was found to be rich in manoalide monoacetate (10) and was not further purified. The fractions containing compounds 1-3 were further purified using semipreparative HPLC [Phenomenex, Luna C18 (2), 5  $\mu$ m, 250  $\times$  10 mm; gradient of CH<sub>3</sub>-CN-H<sub>2</sub>O, 73:27 to 81:21, over 15 min as eluent, flow rate 4 mL/min; UV detection at 254 nm] to afford 25-acetoxyluffariellin A (1), 25acetoxyluffariellin B (2), 25-acetoxyseco-manoalide (3), manoalide (4), seco-manoalide (5), and luffariellin A (8). Compounds 4, 5, 8, and 10 were identified by comparison of their NMR data with literature data.1-3,6

**25-Acetoxyluffariellin A (1):** 4.41 mg, 0.08% dry weight; colorless oil;  $[\alpha]^{21}_{D} - 38.1$  (*c* 0.11, CHCl<sub>3</sub>); UV (PDA, CH<sub>3</sub>CN-H<sub>2</sub>O, 70:30 to 100:0)  $\lambda_{max}$  (relative absorption) 196 (1), 228 (0.67) nm; IR (film)  $\nu_{max}$  3490 (br), 1797, 1766, 1211, 1026, 999 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR, see Tables 1 and 2; HRESIMS *m*/*z* 481.2577 (C<sub>27</sub>H<sub>38</sub>O<sub>6</sub>Na [M + Na]<sup>+</sup> required 481.2561).

**25-Acetoxyluffariellin B (2):** 1.14 mg, 0.02% dry weight; colorless oil;  $[\alpha]^{21}_{D} - 156.1$  (*c* 0.06, CHCl<sub>3</sub>); UV (PDA, CH<sub>3</sub>CN-H<sub>2</sub>O, 70:30 to 100:0)  $\lambda_{max}$  (relative absorption) 200 (1), 226 (0.84) nm; IR (film)  $\nu_{max}$  3518 (br), 2362, 2335, 1761, 1679, 1210 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR, see Tables 1 and 2; HRESIMS *m*/*z* 481.2547 (C<sub>27</sub>H<sub>38</sub>O<sub>6</sub>Na [M + Na]<sup>+</sup> required 481.2561).

**25-Acetoxyseco-manoalide (3):** 3.36 mg, 0.06% dry weight; colorless oil;  $[\alpha]^{21}_{D}$  –42.0 (*c* 0.11, CHCl<sub>3</sub>); UV (PDA, CH<sub>3</sub>CN–H<sub>2</sub>O,

70:30 to 100:0)  $\lambda_{\text{max}}$  (relative absorption) 193 (1), 228 (0.70) nm; IR (film)  $\nu_{\text{max}}$  3480 (br), 1799, 1681, 1208 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR, see Tables 1 and 2; HRESIMS *m*/*z* 481.2556 (C<sub>27</sub>H<sub>38</sub>O<sub>6</sub>Na [M + Na]<sup>+</sup> required 481.2561).

**Analyses of Sponge Extracts.** Each sponge sample (n = 15) was divided into two portions immediately after collection under water. Both portions of each sponge were placed separately in liquid N<sub>2</sub> on return to the surface. In the laboratory, one portion of each sponge was freeze-dried. The remaining portion of each sponge was allowed to thaw at RT for 3 h before being freeze-dried. All freeze-dried sponge samples were extracted with CH<sub>2</sub>Cl<sub>2</sub> ( $3 \times 10$  mL). The combined dried extracts were dissolved in DMSO and analyzed using analytical HPLC [Phenomenex, Luna C18 (2), 5  $\mu$ m, 250 × 4.6 mm; gradient of CH<sub>3</sub>-CN-H<sub>2</sub>O, 73:27 + 0.1% TFA to 81:21 + 0.1% TFA, over 35 min as eluent, flow rate 1 mL/min; UV detection at 254 nm].

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